

**PROCESSES FOR MAKING PHARMACEUTICAL ORAL ECB
FORMULATIONS AND COMPOSITIONS**

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FIELD OF THE INVENTION

- 5 The present invention relates to processes for making pharmaceutical formulations or compositions containing an antifungal compound, in particular, the incorporation of an echinocandin/carbohydrate complex into an oral formulation or composition to enhance stability.

BACKGROUND OF THE INVENTION

- 10 Oral formulations of pharmaceutical drugs may be administered to patients via sachets, chewable or conventional tablets, capsules or oral solutions and suspensions. The formulation that is developed for a particular drug is dependent on a variety of issues. For example, it is well-known in the art that an oral formulation should be sufficiently stable, have an acceptable appearance and be prepared from generally
15 accepted, safe excipients.

- Stability is an important consideration when designing a formulation, especially for oral applications. For practical and commercial reasons, it is desirable to maintain sufficient formulation stability for at least two years. Therefore, a primary formulation design goal is to optimize shelf-life and stability at the selected storage
20 condition (e.g., room temperature).

- The instability of the echinocandin compounds make them particularly difficult to formulate. Most of the formulations tested to date have a shelf life of less than one year at room temperature. Generally, a shelf life of at least two years is desirable. Therefore, there is a need for an oral formulation containing an
25 echinocandin compound having improved thermal stability.

SUMMARY OF THE INVENTION

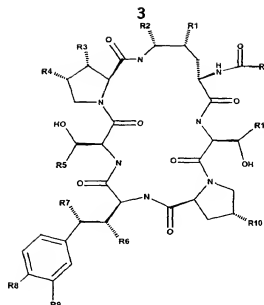
 Applicants have discovered that the incorporation of one or more carbohydrates into an echinocandin formulation at specific process stages provides a significant improvement in thermal stability. In one embodiment of the present

invention, a process is provided for preparing an oral pharmaceutical formulation which comprises (i) mixing an echinocandin compound or a echinocandin/carbohydrate complex and at least one carbohydrate in a solvent or mixture of solvents to form a pharmaceutical solution; (ii) spraying the solution onto a layer of fluidized granular diluent or carrier (e.g., a granular carbohydrate); and (iii) removing the excess solvent or solvents. The pharmaceutical solution may optionally contain surfactants, flavorings, colorants and/or processing aids. Oral granular pharmaceutical formulations prepared by the process described above are also provided, as well as medicaments prepared therefrom such as sachets and chewable tablets.

In another embodiment of the present invention, a process is provided for preparing an oral pharmaceutical formulation comprising the steps of: (i) mixing an echinocandin compound or echinocandin/carbohydrate complex, at least one carbohydrate, and a soluble granulating agent (e.g., polyvinylpyrrolidone) in a solvent or mixture of solvents to form a pharmaceutical solution; (ii) spraying the solution onto a layer of fluidized non-granular diluent or carrier (e.g., carbohydrate powders); and (iii) removing the excess solvent or solvents. The pharmaceutical solution may optionally contain surfactants, flavorings, colorants and/or processing aids. Oral granular pharmaceutical formulations prepared by the process described above are also provided, as well as medicaments prepared therefrom such as sachets and chewable tablets.

In yet another embodiment of the present invention, a method is provided for treating an antifungal infection in a mammal in need thereof which comprises administering an oral formulation prepared by one of the processes described above to the mammal.

The term "echinocandin" refers to a compound having the following general structure:



wherein:

R is an alkyl group, an alkenyl group, an alkynyl group, an aryl group, heteroaryl group, or combinations thereof;

R₁, R₂, R₃, R₆, R₇, and R₁₀ are independently hydroxy or hydrogen;

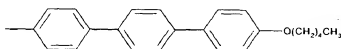
R₄ is hydrogen, methyl or -CH₂C(O)NH₂;

R₅ and R₁₁ are independently methyl or hydrogen;

R₈ is -OH, -OPO₃H₂, -OPO₃HCH₃, -OPO₂HCH₃, or -OSO₃H; and

R₉ is -H, -OH, or -OSO₃H.

"Echinocandin B" or "ECB" refers to a echinocandin compound as described above where R₁, R₂, R₃, R₆, R₇, R₈ and R₁₀ are hydroxy groups; R₄, R₅ and R₁₁ are methyl groups; R₉ is a hydrogen. In the natural product, R is a linoleoyl group. In a particularly useful semi-synthetic compound, R has both a rigid and a flexible component, for example, a compound where R is represented by the following



The term "alkyl" refers to a hydrocarbon radical of the general formula C_nH_{2n+1} , containing from 1 to 30 carbon atoms unless otherwise indicated. The alkane radical may be straight, branched, cyclic, or multi-cyclic. The alkane radical may be substituted or unsubstituted. Similarly, the alkyl portion of an alkoxy group or alkanooate have the same definition as above.

The term "alkenyl" refers to an acyclic hydrocarbon containing at least one carbon-carbon double bond. The alkene radical may be straight, branched, cyclic, or multi-cyclic. The alkene radical may be substituted or unsubstituted.

The term "alkynyl" refers to an acyclic hydrocarbon containing at least one carbon carbon triple bond. The alkyne radical may be straight, or branched. The alkyne radical may be substituted or unsubstituted.

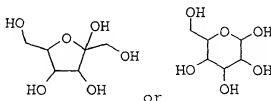
The term "aryl" refers to aromatic moieties having single (e.g., phenyl) or fused ring systems (e.g., naphthalene, anthracene, phenanthrene, etc.). The aryl groups may be substituted or unsubstituted. Substituted aryl groups include a chain of aromatic moieties (e.g., biphenyl, terphenyl, phenylnaphthyl, etc.)

The term "heteroaryl" refers to aromatic moieties containing at least one heteroatom within the aromatic ring system (e.g., pyrrole, pyridine, indole, thiophene, furan, benzofuran, imidazole, pyrimidine, purine, benzimidazole, quinoline, etc.). The aromatic moiety may consist of a single or fused ring system. The heteroaryl groups may be substituted or unsubstituted.

"Echinocandin/carbohydrate complex" refers to a crystalline complex formed between an echinocandin compound and a carbohydrate when the echinocandin is crystallized or recrystallized from a solvent in the presence of the carbohydrate. A more detailed description of the echinocandin/carbohydrate complexes may be found in Larew, et al., filed on March 3, 1999 entitled "Echinocandin/Carbohydrate Complexes" and incorporated herein by reference.

"Carbohydrate" refers to an aldehydic or ketonic derivative of polyhydric alcohols represented by the formulas $C_n(H_2O)_n$ (e.g., glucose, $C_6(H_2O)_6$; sucrose, $C_{12}(H_2O)_{11}$). Carbohydrates include compounds with relatively small molecules, such

as the simple sugars (e.g., monosaccharides, disaccharides, etc.), as well as macromolecular (polymeric) substances such as starch, glycogen, and cellulose polysaccharides. Sugars are carbohydrates (saccharides) having the general composition $(\text{CH}_2\text{O})_n$ and simple derivatives thereof. Although the simple monomeric sugars (glycoses) are described as polyhydroxy aldehydes or ketones, e.g., $\text{HOCH}_2-(\text{CHOH})_4-\text{CHO}$ for aldohexoses (e.g., glucose) or $\text{HOCH}_2-(\text{CHOH})_3-\text{CO}-\text{CH}_2\text{OH}$ for 2-ketoses (e.g., fructose), the structures are commonly written as five (furanose) or six (pyranose) membered ring cyclic ethers, e.g.



The term "granular" diluents or carriers refer to materials where approximately 70% by weight of the material has a particle size greater than 150 microns.

The term "non-granular" or "powder" diluents or carriers refer to materials where approximately 70% by weight of the material has a particle size less than 150 microns.

The term "granulating agent" refers to a binder that causes the diluent or carrier to bind together to form larger aggregates or granules upon spraying the agent onto the diluent powder surface in the presence of a solvent or solvents.

DETAILED DESCRIPTION

Applicants have discovered that the stability of an oral echinocandin formulation or composition may be enhanced by incorporating the echinocandin compound into the pharmaceutical composition as an echinocandin/carbohydrate complex or by forming an echinocandin/carbohydrate complex *in situ*. For example, the echinocandin/carbohydrate complex may be formed *in situ* by mixing the echinocandin compound with the carbohydrate in a suitable solvent to provide a pharmaceutical solution which is then sprayed onto a fluidized bed of granules. Alternatively, the echinocandin/carbohydrate pharmaceutical solution may be sprayed

onto a fluidized bed of non-granular (or powder) diluent or carrier. Preferably, a granulating agent is added to the spray solution to bind the powder together to form larger aggregates or granules in the presence of the solvent or solvents. To perform most effectively, the granulating agent should be soluble in the solvent system used in the pharmaceutical solution. For an acetone/water solvent system, a preferred granulating agent is polyvinylpyrrolidone (e.g., Povidone™ available from BASF Corp.).

Suitable carbohydrates include adonitol, arabinose, arabitol, ascorbic acid, chitin, D-cellulose, 2-deoxy-D-ribose, dulcitol, (S)-(+)-erythrose, fructose, fucose, galactose, glucose, inositol, lactose, lactulose, lyxose, maltitol, maltose, maltotriose, mannitol, mannose, melezitose, melibiose, microcrystalline cellulose, palatinose, pentaerythritol, raffinose, rhamnose, ribose, sorbitol, sorbose, starch, sucrose, trehalose, xylitol, xylose and hydrates thereof. Suitable carbohydrates also include the D and L enantiomers, as well as the alpha and beta anomers of the compounds listed above. Preferred carbohydrates are the simple sugars (e.g., mono- and disaccharides). Particularly useful sugars include L-arabinose, D-arabitol, L-arabitol, 2-deoxy-D-ribose, (S)-(+)-erythrose, D-fructose, D-(+)-fucose, L-fucose, D-galactose, β-D-glucose, D-lyxose, L-lyxose, D-maltose, maltotriose, melezitose, palatinose, D-raffinose, D-sorbitol, D-trehalose, xylitol, L-xylose and hydrates thereof. When an echinocandin/carbohydrate complex is used, the added carbohydrate may or may not be the same as the carbohydrate in the complex.

Suitable solvents include any solvent, or mixture of solvents, inert to the ongoing process that sufficiently solubilizes the echinocandin and carbohydrate materials to afford a medium which can be used in the fluid bed spray process (typically, protic or ketone solvents). A preferred solvent system used for the pharmaceutical spray solution is a mixture of water and acetone. While any percent mixture of water and acetone can be used that will solubilize the echinocandin or echinocandin/carbohydrate complex and carbohydrate, the preferred percentage of acetone is from about 50% to 70% on a volume basis.

Suitable granular and non-granular (or powder) materials include the carbohydrates described above as well as cellulose/starch materials, polyethylene glycol 4000, 6000 and 8000 (e.g., the higher melting solid PEG series; available from Union Carbide Corp.), hydroxypropyl methylcellulose (HPMC; available from Shin-
5 Etsu Chemical Co. Ltd.), hydroxypropyl methylcellulose phthalates (e.g., HP-50, HP-55, HP-55S; also available from Shin-Etsu Chemical Co. Ltd), and dextrates (e.g., Emdex™; available from Edward Mendell Co. Inc. and Dextrin™; available from Matheson Coleman & Bell). Preferred granular or non-granular materials are the carbohydrates such as fructose, glucose, lactose, lactulose, maltitol, maltose,
10 maltotriose, mannitol, mannose, microcrystalline cellulose, HPMC, HPMC Phthalate, dextrates, dextrin, sorbitol, sorbose, starch and starch derivatives, sucrose, trehalose, xylitol, xylose and hydrates thereof. Suitable carbohydrates also include the D and L enantiomers, as well as the alpha and beta anomers. Most preferred are the low moisture mono- and di-saccharides, such as mannitol (e.g., Mannogen™ 2080,
15 available from SPI Polyols, Inc., New Castle, DE), lactose and maltose.

The pharmaceutical solution is sprayed onto the fluidized granular or non-granular carrier in a fluid bed granulator/dryer (e.g., Glatt GPCG-3, available from Glatt Air Techniques, Inc., Ramsey, New Jersey). Although not wishing to be bound by any particular theory, it is believed that the sprayed pharmaceutical solution forms
20 an *in situ* echinocandin/carbohydrate complex on the surface of the fluidized carrier upon drying. The echinocandin compound is generally present in the final sprayed granules at a weight percentage from about 5% to 25%, preferably from 7% to 20%, more preferably from 12% to 16%. The carbohydrate used in the pharmaceutical solution is generally present in the final sprayed granules at a weight percentage from
25 about 5% to 25%, preferably from about 7% to 20%, more preferably from about 12% to 16%. The granular or non-granular carrier is generally present in the final sprayed granules at a weight percentage from about 50% to 90%, preferably from about 60% to 80%, more preferably from about 65% to 75%.

Applicants have observed that the final sprayed granules exhibit better thermal
30 stability and enhanced flowability compared to the starting bulk echinocandin material alone. (see, Example 1) Even though one of the major excipients of the formulation

(i.e., fluidized carbohydrate - a potential echinocandin stabilizer) is not part of the solution containing the pharmaceutically active compound, the sprayed granules still exhibit enhanced stability.

The relative thermal storage stability of the fluid bed spray granules was assessed and compared to that of the starting echinocandin bulk. HPLC analysis of the starting echinocandin bulk and of the final spray granules was run to determine the initial level of degradation products, often referred to as % total related substances (i.e. % TRS). Both samples were then stored in separate high density polyethylene (HDPE) bottles with wax vinyl aluminum foil (WVAF) cap seals for 1 and 2 weeks at 40°C. The increase in % TRS was measured by HPLC at the two time points. Table 1 shows the change in %TRS between bulk ECB compound and a trial process for ECB granules prepared by the fluid bed spray process described above.

Table 1

Storage Conditions*	Starting ECB bulk Compound (%TRS)	FB Spray Granules (%TRS)
Initial	4.94%	5.12%
1 Week at 40°C	+ 2.86%	+ 1.38%
2 Weeks at 40°C	+ 3.85%	+ 2.30%

* Each of the samples were stored in brown opaque HDPE bottles closed with WVAF caps.

The FB spray process did not significantly increase the %TRS of the sprayed granules. It is important to note that the FB spray granules containing the *in-situ* formed echinocandin/carbohydrate complex showed enhanced storage stability (i.e. lower % TRS increases) at 1 and 2 weeks of storage at 40°C relative to the starting echinocandin bulk compound.

Table 2 shows the change in %TRS between bulk ECB compound and a second trial process for ECB granules prepared by the fluid bed spray process described above.

Table 2

Storage Conditions*	Starting ECB bulk Compound (%TRS)	FB Spray Granules (%TRS)
Initial	5.36%	5.64%
1 Week at 40°C	+3.46%	+1.10%
2 Weeks at 40°C	+4.29%	+2.51%

* Each of the samples were stored in brown opaque HDPE bottles closed with WVAF caps.

5 Again, the FB spray process did not significantly increase the % TRS of the sprayed granules. The FB spray granules containing the *in-situ* formed echinocandin/carbohydrate complex showed enhanced storage stability (i.e. lower % TRS increases) at 1 and 2 weeks of storage at 40°C relative to the starting echinocandin bulk compound. In addition to the enhanced stability of the FB spray
10 granules, improved mixability and flowability of the final FB spray granules was observed.

The sprayed granules may be used to prepare a pharmaceutical medicament that can be orally administered to a patient in need of therapeutic treatment for which the product is designed. For example, the sprayed granules may be contained in a
15 sachet or capsule, or compressed into a tablet. The final spray granules can be readily filled directly into sachet packages, or alternatively, can be readily mixed with additional excipients prior to forming into tablets, filling into capsules or filling into bottles.

For example, the sprayed granules can be dry mixed with flavoring agent(s),
20 tablet disintegrant (e.g. Croscarmellose sodium; available from FMC Corp.), processing aid(s) (e.g. Syloid 63FP; silicon dioxide available from Grace-Davison, Baltimore MD) and tablet lubricants such as stearic acid (available from Witco Corp.) or glyceryl behenate (available from Gattefosse Corp.) prior to being compressed into tablets (chewable) using standard tableting equipment and technology. (see i.e.,
25 "Oral Solid Dosage Forms", Remington's Pharmaceutical Sciences, Chapter 90, Mack Publishing Co., Pennsylvania, 1985.)

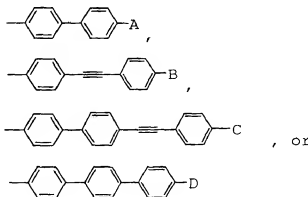
Chewable tablets may be produced using standard procedures well known to those skilled in the art. Generally, chewable tablets contain flavoring agents (e.g. peppermint oil, menthol, or wintergreen oil, etc.) at levels between about 0.1% to 2.0% based on total final tablet weight (preferably between about 0.2% and 0.6% by weight) which may be added to the echinocandin/carbohydrate pharmaceutical solution prior to spraying the granules. Sweeteners are also generally added to chewable tablets. However, in the present invention, the carbohydrates used in the formulation may provide sufficient sweetening such that additional sweeteners may not be necessary. The use of mannitol or maltose as the carrier in the formulation would also supply additional sweetening plus the proper "mouth-feel"/"coolness" and texture for chewable tablets. Other ingredients may also be added by dry mixing (e.g. tumble bin or ribbon mixer) with the sprayed granules to give the final granulation mix. The final granulation mix is then compressed into tablets. Unlike conventional tablets, larger tooling (e.g. Pillow shaped No. 800 tooling) is generally used in the production of chewable tablets.

In addition to the materials already discussed, the oral formulations or medicaments of the present invention may also include other carriers, diluents and excipients which are well known to those skilled in the art and include materials such as waxes, water soluble and/or swellable polymers, hydrophilic or hydrophobic materials, gelatin, oils, solvents, water, and the like. The particular carrier, diluent or excipient used will depend upon the means and purpose for which the active ingredient is being applied. The formulations may also include wetting agents, lubricating agents, emulsifiers, suspending agents, preservatives, perfuming agents, colorants, opaquing agents, glidants, processing aids and combinations thereof.

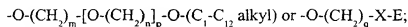
The cyclic peptides used in the present invention may be produced by culturing various microorganisms. Suitable natural product starting materials belonging to the echinocandin cyclic peptide family include Echinocandin B, Echinocandin C, Echinocandin D, Aculeacin A_y, Mulundocandin, Sporiofungin A, Pneumocandin A_o, WF11899A, and Pneumocandin B_o. In general, the cyclic peptides may be characterized as a cyclic hexapeptide nucleus with an acylated amino group on one of the amino acids. The amino group on the naturally-occurring cyclic peptide is

typically acylated with a fatty acid group forming a side chain off the nucleus. Examples of naturally-occurring acyl groups include linoleoyl (Echinocandins B, C and D), palmitoyl (Aculeacin A and WF11899A), stearoyl, 12-methylmyristoyl (Mulundocandin), 10,12-dimethylmyristoyl (Sporiofungin A and Pneumocandin A₀) and the like.

Semi-synthetic derivatives may be prepared by removing the fatty acid side chain from the cyclic peptide nucleus to produce a free amino group (i.e., no pendant acyl group -C(O)R). The free amine is then reacylated with a suitable acyl group. For example, the echinocandins B nucleus has been re-acylated with certain non-naturally occurring side chain moieties to provide a number of antifungal agents. See, i.e., U.S. Patent No. 4,293,489 (Debono). Those skilled in the art will appreciate that the N-acyl side chain encompasses a variety of side chain moieties known in the art. Suitable side chain moieties include substituted and unsubstituted alkyl groups, alkenyl groups, alkynyl groups, aryl groups, heteroaryl groups and combinations thereof. Preferably, the side chain contains both a linearly rigid section and a flexible alkyl section to maximize antifungal potency. Representative examples of preferred acyl side chains include R groups having the following structures:



where A, B, C and D are independently hydrogen, C₁-C₁₂ alkyl, C₂-C₁₂ alkynyl, C₁-C₁₂ alkoxy, C₁-C₁₂ alkylthio, halo, or



m is 2, 3 or 4; n is 2, 3 or 4; p is 0 or 1; q is 2, 3 or 4;

X is pyrrolidino, piperidino or piperazino; and E is hydrogen, C₁-C₁₂ alkyl, C₃-C₁₂ cycloalkyl, benzyl or C₃-C₁₂ cycloalkylmethyl.

As noted above, the cyclic peptides described herein may be prepared by fermentation of known microorganisms as described in the art. The subsequent deacylation is typically carried out enzymatically using a deacylase enzyme by known materials and procedures described in the art.

For example, U.S. Patent No. 3,293,482 (Abbott et al.), incorporated herein by reference, describes the deacylation and preparation of the cyclic peptide of formula I where R₄, R₅, and R₁₁ are methyl, R₆ is hydrogen, and R₁, R₂, R₃, R₆, R₇, R₈ and R₁₀ are each hydroxy. U.S. Patent No. 4,299,763 (Abbott et al.), incorporated herein by reference, describes the deacylation and preparation of the cyclic peptide of formula I where R₄, R₅, and R₁₁ are methyl, R₂ is hydroxy, R₇ and R₉ are hydrogen and R₁, R₃, R₆, R₈ and R₁₀ are each hydroxy. U.S. Patent No. 3,978,210 (Mizuno et al.), incorporated herein by reference, describes the preparation of aculeacin. U.S. Patent No. 4,304,716, incorporated herein by reference, describes the deacylation and preparation of the cyclic peptide of formula I where R₅ is -CH₂C(O)NH₂; R₁₁ is methyl; R₄ and R₉ are hydrogen; R₁, R₂, R₃, R₆, R₇, R₈ and R₁₀ are each hydroxy and the acyl group with substituent R is myristoyl.

Cyclic peptides where R₂ and R₇ are each hydrogen may be prepared by subjecting the corresponding compound (where R₂ and R₇ are each hydroxy; the ornithine alpha-amino group may be a free amino group or acylated) to a strong acid and a reducing agent at a temperature of between -5°C and 70°C, in a suitable solvent. Suitable strong acids include trichloroacetic acid, trifluoroacetic acid or boron trifluoride etherate. A preferred strong acid is trifluoroacetic acid. Suitable reducing agents include sodium cyanoborohydride or triethylsilane. A preferred reducing agent is triethylsilane. Suitable solvents include methylene chloride, chloroform or acetic acid, preferably methylene chloride. The strong acid is present in an amount from about 2 to 60 mol per mol of substrate, and the reducing agent is present in an amount from about 2 to 60 mol per mol of substrate. The acid reduction process selectively removes the aminor (R₂) and benzylic (R₇) hydroxy groups.

Acylation of the α -amino group on the ornithine unit may be accomplished in a variety of ways well known to those skilled in the art. For example, the amino group may be acylated by reaction with an appropriately substituted acyl halide, preferably in the presence of an acid scavenger such as a tertiary amine (e.g., triethylamine). The reaction is typically carried out at a temperature between about 20°C to 25°C. Suitable reaction solvents include polar aprotic solvents, such as dioxane or dimethylformamide. Solvent choice is not critical so long as the solvent employed is inert to the ongoing reaction and the reactants are sufficiently solubilized to effect the desired reaction.

The amino group may also be acylated by reaction with an appropriately substituted carboxylic acid, in the presence of a coupling agent. Suitable coupling agents include dicyclohexylcarbodiimide (DCC), N,N'-carbonyldiimidazole, bis(2-oxo-3-oxazolidinyl)phosphinic chloride (BOP-Cl), N-ethoxycarbonyl-2-ethoxy-1,2-dihydroquinoline (EEDQ), benzotriazole-1-yloxy-tripyrrolidinophosphonium hexafluorophosphate (PyBOP) and the like.

Alternately, the amino group may be acylated with an activated ester of a carboxylic acid such as p-nitrophenyl, 2,4,5-trichlorophenyl, hydroxybenzotriazole hydrate (HOBt-H₂O), pentafluorophenol, and N-hydroxysuccinimide carboxylate esters. Preferred acylating moieties are the 2,4,5-trichlorophenyl and HOBt carboxylate esters. The reaction is typically ran 1 to 65 hours at a temperature from about 0°C to 30°C in an aprotic solvent. The reaction is generally complete after about 24 to 48 hours when carried out at a temperature between about 15°C to 30°C. Suitable solvents include tetrahydrofuran and dimethylformamide or mixtures thereof. The amino group is generally present in equimolar proportions relative to the activated ester or with a slight excess of the amino group.

The R-COOH precursor acids are prepared by hydrolyzing a nitrile of the formula R-CN or an ester of the formula R-COO(C₁-C₄ alkyl). The nitrile and ester intermediates may be prepared using procedures known in the art. For example, the nitrile and ester intermediates where R is an alkoxy aryl moiety may be prepared using Procedure A or Procedure B, described below.

Procedure A

One equivalent of an alkyl bromide, iodide, or *p*-toluenesulfonate is added to a mixture containing one equivalent of a base, such as potassium *t*-butoxide or potassium carbonate (K_2CO_3), and one equivalent of an hydroxy aryl compound in 200-300ml of acetonitrile (CH_3CN). The reaction mixture is refluxed for 6h and then concentrated *in vacuo* to provide a residue which is dissolved in a $Et_2O/2N$ NaOH mixture. The resulting layers are separated and the organic layer is dried over magnesium sulfate ($MgSO_4$), filtered and dried to provide the alkoxy aryl product.

Procedure B

Diethylazodicarboxylate (1 equiv.) is added dropwise to a mixture containing an hydroxy aryl compound (1 equiv.), an alkyl alcohol (1 equiv.) and triphenylphosphine (1 equiv.) in 200-300ml of THF. After 17h, the solvent is removed *in vacuo* to provide a residue which is dissolved in Et_2O . The resulting mixture is washed with a 2N NaOH solution, dried over $MgSO_4$, filtered and concentrated to provide a product which is then crystallized from a Et_2O /pentane mixture or, if the product contains a tertiary amine, the hydrochloride salt is formed and crystallized from a methanol ($MeOH$)/ $EtOAc$ mixture. The nitrile and ester intermediates where R is an alkynyl aryl moiety may be prepared using Procedure C, below.

Procedure C

A mixture containing Et_2O (2 equiv.), palladium dichloride (0.05 equiv.), triphenylphosphine (0.1 equiv.), cuprous iodide (0.025 equiv.) and an alkyne (1 equiv.) is added to one equivalent of an aryl bromide, iodide, or trifluoromethanesulfonate in CH_3CN (600ml/0.1 mol of aryl reactant), under nitrogen (N_2). The resulting mixture is refluxed for 17h and then the solvent is removed *in vacuo* to provide a residue which is slurried in 300 ml of Et_2O and then filtered. The filtrate is washed with a 1N HCl solution, dried over $MgSO_4$, filtered and then dried to provide the product.

The ester intermediates where R is a terphenyl moiety may be prepared using Procedure D, below.

Procedure D

1. Formation of boronic acid reactant

5 Butyl lithium (1.2 equivalents) is added to one equivalent of a cold (-78°C) aryl halide in THF. After 15 minutes, triisopropyl borate (2 equiv.) is added. After 10 minutes, the reaction mixture is warmed to room temperature and quenched by the addition of water (H₂O), followed by the addition of 1N HCl. The resulting layers are separated and the organic layer is concentrated *in vacuo* to provide a solid
10 which is collected by filtration and washed with hexane.

2. Formation of terphenyl ester

Tetrakis(triphenylphosphine)palladium (0.03 equiv.) is added to a mixture containing an aryl boronic acid (1 equiv.), K₂CO₃ (1.5 equiv.) and methyl 4-iodobenzoate (1 equiv.) (or trichlorophenyl ester of iodobenzoate) in N₂-purged
15 toluene. The reaction mixture is refluxed for 7h and then decanted to remove the K₂CO₃ and dried *in vacuo* to provide a residue. This residue is triturated in CH₃CN and filtered to provide the product. The aryl nitriles and esters described above may be converted to the corresponding carboxylic acids by hydrolysis using Procedure E or Procedure F, below.

Procedure E

20 An aryl nitrile is dissolved in ethanol (EtOH) and an excess of 50% NaOH solution and refluxed for 2h. Water is added to the reaction mixture until a solid precipitates. This solid is collected by filtration, added to a dioxane/6N HCl mixture and the resulting mixture is refluxed for 17h. When the reaction is substantially
25 complete, the carboxylic acid product is crystallized by the addition of H₂O and then collected by filtration and dried *in vacuo*.

Procedure F

An excess of 2N NaOH is added to an aryl ester in MeOH, and the resulting solution is refluxed for 5h and then acidified by the addition of excess HCl. Water is added to the reaction mixture until a solid (carboxylic acid) precipitates. The carboxylic acid is collected by filtration and dried *in vacuo*.

- 5 The carboxylic acids may be converted to the corresponding 2,4,5-trichlorophenyl esters using Procedure G, below. The activated esters are then used to acylate the amino nucleus.

Procedure G

- 10 A mixture containing an aryl carboxylic acid (1 equiv.), 2,4,5-trichlorophenol (1 equiv.) and DCC (1 equiv.) in CH_2Cl_2 is stirred for 17h and then filtered. The filtrate is concentrated to provide a residue which is dissolved in Et_2O , filtered, and then pentane is added until crystallization begins. The crystals are collected by filtration and dried *in vacuo*.

- 15 Alternatively, the carboxylic acid may be activated by conversion to the corresponding hydroxybenzotriazole ester using Procedure H, below.

Procedure H

- 20 An aryl carboxylic acid (1 equiv.) and a slight excess of N-mesylate substituted hydroxybenzotriazole (1.2 equiv.) were reacted in the presence of a slight excess of a base such as triethylamine (Et_3N) (1.3 equiv.) in DMF, under N_2 . When the reaction was complete, the mixture was diluted with toluene and washed with H_2O . The organic portion was diluted with H_2O and then filtered using *t*-butyl methyl ether (MTBE) for transferring the material. The resultant solid was washed with MTBE and then dried *in vacuo*.

- 25 The echinocandin compound may be isolated and used per se or in the form of its pharmaceutically acceptable salt or hydrate. The term "pharmaceutically acceptable salt" refers to non-toxic acid addition salts derived from inorganic and organic acids. Suitable salt derivatives include halides, thiocyanates, sulfates, bisulfates, sulfites, bisulfites, arylsulfonates, alkylsulfates, phosphonates, monohydrogen-phosphates, dihydrogenphosphates, metaphosphates,

pyrophosphonates, alkanoates, cycloalkylalkanoates, arylalkonates, adipates, alginates, aspartates, benzoates, fumarates, glucoheptanoates, glycerophosphates, lactates, maleates, nicotines, oxalates, palmitates, pectinates, picrates, pivalates, succinates, tartarates, citrates, camphorates, camphorsulfonates, digluconates, trifluoroacetates, and the like.

Alternatively, the echinocandin compound may be isolated as a Echinocandin/carbohydrate complex. The complexes are formed using standard crystallization procedures such as those typically performed for purifying compounds by recrystallization. The echinocandin material and carbohydrate are dissolved at an elevated temperature (approximately 45 to 60°C, preferably < than 55°C) in a solvent. The solution is then slowly cooled until the crystallization begins. A seed crystal (such as a previously crystallized complex or an insoluble sugar) may be added to initiate crystallization. Suitable carbohydrates include those listed earlier for the carbohydrates added to the pharmaceutical spray solution. Preferred carbohydrates include L-arabinose, D-arabitol, L-arabitol, 2-deoxy-D-ribose, (S)-(+)-erythrose, D-fructose, D-(+)-fucose, L-fucose, D-galactose, β -D-glucose, D-lyxose, L-lyxose, D-maltose, maltotriose, melezitose, palatinose, D-raffinose, D-sorbitol, D-trehalose, xylitol, L-xylitol and hydrates thereof. Suitable solvents include any solvent, or mixture of solvents, inert to the ongoing reaction that sufficiently solubilizes the reactants to afford a medium within which to effect the desired complexation between the carbohydrate and the echinocandin compound, such as protic or ketone solvents including methanol, ethanol, benzyl alcohol, as well as mixtures of benzyl alcohol with solvents such as methanol, ethanol, *n*-propanol, isopropanol, *n*-butanol, 2-butanol, *t*-butanol, 2-pentanol, 2-methyl-1-propanol, MEK, acetone, ethyl acetate, toluene, acetonitrile, fluorobenzene, methylene chloride, nitromethane, or cyclic ketones such as cyclopentanone and cyclohexanone. Preferred solvents include methanol, ethanol, benzyl alcohol, and mixtures of benzyl alcohol with methyl ethyl ketone, ethyl acetate, and acetonitrile. A more detailed description of echinocandin/carbohydrate complexes and their preparation may be found in Larew, et al., filed on March 3, 1999 entitled "Echinocandin/Carbohydrate Complexes" and incorporated herein by reference.

The active ingredient is typically formulated into pharmaceutical dosage forms to provide an easily controllable dosage of the drug and to give the patient an elegant and easily handleable product. As used herein, the term "unit dose" or "unit dosage" refers to physically discrete units that contain a predetermined quantity of active ingredient calculated to produce a desired therapeutic effect. When a unit dose is administered orally, it is typically provided in the form of a tablet, capsule, pill, powder packet (sachet), wafer, etc. Alternatively, a unit dose may be administered in the form of a liquid solution that can be swished and swallowed, or a dry or liquid aerosol which may be inhaled or sprayed.

The dosage to be administered may vary depending upon the physical characteristics of the patient, the severity of the patient's symptoms, and the means used to administer the drug. The specific dose for a given patient is usually set by the judgment of the attending physician.

Echinocandin and semi-synthetic echinocandin compounds have been shown to exhibit antifungal and antiparasitic activity such as growth inhibition of various infectious fungi including *Candida* spp. (i.e., *C. Albicans*, *C. Parapsilosis*, *C. Krusei*, *C. Glabrata*, *C. Tropicalis*, or *C. Lusitaniaw*); *Torulopus* spp. (i.e., *T. Glabrata*); *Aspergillus* spp. (i.e., *A. Fumigatus*); *Histoplasma* spp. (i.e., *H. Capsulatum*); *Cryptococcus* spp. (i.e., *C. Neoformans*); *Blastomyces* spp. (i.e., *B. Dermatitidis*); *Fusarium* spp.; *Trichophyton* spp., *Pseudallescheria boydii*, *Coccidioides immitis*, *Sporothrix schenckii*, etc.

Compounds of this type also inhibit the growth of certain organisms primarily responsible for opportunistic infections in immunosuppressed individuals, such as growth inhibition of *Pneumocystis carinii* (the causative organism of pneumocystis pneumonia (PCP) in AIDS and other immunocompromised patients. Other protozoans that are inhibited by echinocandin-type compounds include *Plasmodium* spp., *Leishmania* spp., *Trypanosoma* spp., *Cryptosporidium* spp., *Isospora* spp., *Cyclospora* spp., *Trichomonas* spp., *Microsporidiosis* spp., etc.

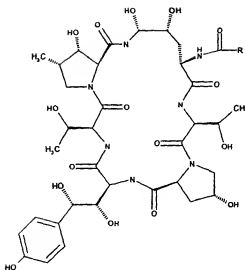
Consequently, the formulations generated are useful in combating either systemic fungal infections or fungal skin infections. Accordingly, the formulations

and processes of the present invention may be used in the manufacture of a medicament for the therapeutic applications described herein. For example, fungal activity (preferably, *Candida albicans* or *Aspergillus fumigatis* activity) or parasitic activity may be inhibited by contacting a pharmaceutical formulation prepared by the present invention with a fungus or parasite, respectively. The term "contacting" includes a union or junction, or apparent touching or mutual tangency of a compound of the invention with a parasite or fungus. The term does not imply any further limitations to the process, such as by mechanism of inhibition. The methods are defined to encompass the inhibition of parasitic and fungal activity by the action of the compounds and their inherent antiparasitic and antifungal properties.

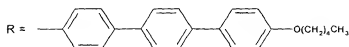
A method for treating a fungal infection which comprises administering an effective amount of a pharmaceutical formulation prepared by the present invention to a host in need of such treatment is also provided. A preferred method includes treating a *Candida albicans* or *Aspergillus fumigatis* infection. The term "effective amount" refers to an amount of active compound which is capable of inhibiting fungal activity. The dose administered will vary depending on such factors as the nature and severity of the infection, the age and general health of the host and the tolerance of the host to the antifungal agent. The particular dose regimen likewise may vary according to these factors. The medicament may be given in a single daily dose or in multiple doses during the day. The regimen may last from about 2-3 days to about 2-3 weeks or longer. A typical daily dose (administered in single or divided doses) contains a dosage level between about 0.01 mg/kg to 100 mg/kg of body weight of an active compound. Preferred daily doses are generally between about 0.1 mg/kg to 60 mg/kg and more preferably between about 2.5 mg/kg to 40 mg/kg.

EXAMPLES

The echinocandin compound used to exemplify the formulations of the present invention was prepared as described in the following preparations. Specifically, the following sequence describes the preparation of anti-fungal compound 1(a) having the following structure:



1 (a)



It will be understood by those skilled in the art that the following serves as an illustrative example and that other semi-synthetic echinocandin compounds useful as anti-fungal agents may be synthesized using similar procedures or procedures described in references cited earlier in the specification. Materials used in the following preparations are available from Aldrich Chemicals (Milwaukee, Wisconsin) unless designated otherwise.

Compound Preparations

Preparation of 4-Bromo-4'-pentyloxybiphenyl 1(a):

Anhydrous K_2CO_3 (416g, 3mol) was added to a mixture of 4-bromo-4'-hydroxybiphenyl (300g, 1.2mol), 1-iodopentane (234ml, 1.79mol) and 2-butanone (600ml). The reaction mixture was refluxed for 44h until TLC (85:15 hexanes/EtOAc) showed complete consumption of the bromo alcohol. The mixture was cooled to about 30°C, diluted with CH_2Cl_2 (600ml) and then filtered. The filtrate was washed twice with H_2O and twice with a saturated aqueous NaCl solution, dried over anhydrous Na_2SO_4 , filtered and then dried at reduced pressure to provide a solid. This solid was isolated by filtration, washed repeatedly with a total of 2L of ice-cold

heptane to remove all traces of iodopentane and then dried overnight under high vacuum. Yield: 340g (88%) of a white powder.

Alternative Preparation of 4-bromo-4'-pentyloxybiphenyl 1(a):

4-Bromo-4'-hydroxybiphenyl (12.5g, 50.2mmol) was added to a solution of
5 NaOH (2.28g, 97% pure, 55.2mmol) in deionized H₂O (150ml), followed by the
addition of 1-iodopentane (11.9g, 60.2mmol) and tetrabutylammonium bromide
(0.82g, 2.51mmol). The mixture was stirred at 90°C for 3.75h until the solids went
into solution. Then, as the reaction proceeded, the desired product began to
precipitate. The mixture was slowly cooled and then filtered to provide a solid which
10 was washed with deionized water until the pH of the filtrate was neutral and then
dried for 16h in a vacuum oven at 30°C. Yield: 15.41g (96%) of 5a. R_f 0.5 (97:3
hexanes/EtOAc). ¹H NMR: δ 0.93 (t, 3H, J=6.9Hz); 1.41 (m, 4H); 1.79 (m, 2H); 3.97
(t, 2H, J= 6.6Hz); 6.98 (m, 2H); 7.23 (m, 6H). ¹³C NMR: δ 14.03; 22.43; 28.22;
28.98; 68.12; 114.91; 120.71; 127.93; 128.27; 131.77; 132.24; 139.82; 159.03.
15 MS(FAB⁺): m/z 320. IR(CHCl₃): 2960, 2936, 2874, 1608, 1518, 1485, 1475 cm⁻¹.
Analysis for C₁₁H₁₃BrO: Calcd: C, 63.96; H, 6.00; Br, 25.0; Found: C, 64.10; H,
5.97; Br, 25.28.

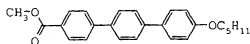
Preparation of 4-Boronic acid-4'-pentyloxybiphenyl 2(a):

To a cold (-20°C) mixture of Compound 1(a) (100g, 0.31 mol) in *t*-
20 butylmethylether (MTBE) (1L), was slowly added *n*-butyl lithium (150ml of a 2.5M
hexanes solution, 0.37mol) dropwise under N₂, while maintaining the internal
temperature between -19° and -18°C. The resultant mixture was stirred for 3.5h
between -17° and -16°C which resulted in a light yellow-green solution. This solution
was cooled to -78°C and diluted with 100ml of anhydrous THF which resulted in a
25 white precipitate. Then, a cold (-78°C) solution of triisopropylborate (145ml,
0.62mol) in MTBE (200ml), under nitrogen was added dropwise over 1.5h while
maintaining the reaction temperature between -78° and -74°C. The resultant reaction
mixture was stirred for 1.5h at -78°C, then allowed to warm to -50°C over 1h at which
time the cooling bath was removed and the mixture was stirred overnight (16-21h)
30 which resulted in a white precipitate. The mixture was shaken vigorously with 2M

HCl (1000ml) for 5 minutes and then the resulting layers were separated and the organic layer was dried at reduced pressure to provide a residue. This residue was diluted with MTBE (100ml), followed by heptane (800ml) to provide a white powder which isolated by suction filtration and washed 3 times with heptane (300ml).

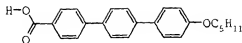
- 5 Yield: 88g (98%). R_f 0.45 (95:5 $\text{CH}_2\text{Cl}_2/\text{MeOH}$). ^1H NMR: δ 0.92 (m, 3H); 1.41 (m, 4H); 1.80 (m, 2H); 4.00 (m, 2H); 6.99 (m, 2H); 7.45-7.63 (m, 3H); 7.67 (m, 2H); 8.24 (d, 1H, $J=8.3\text{Hz}$). ^{13}C NMR: 14.01; 22.26; 28.03; 28.77; 39.61; 39.89; 40.17; 40.45; 67.82; 114.77; 125.32; 127.83; 132.93; 134.84; 141.88; 158.71. MS(FD $^+$): m/z 284. IR(CHCl_3): 2959, 2952, 2874, 1606, 1526, 1500 cm^{-1} .

10 Preparation of Compound 3(a):



3 (a)

- A solution of toluene (174ml) and propanol (20ml) was degassed 3 times by applying vacuum to the solution for 20-30 seconds followed by purging with N_2 . A
 15 2M solution of Na_2CO_3 was also degassed. The toluene/propanol solution (97ml) was added to a mixture of methyl 4-iodobenzoate (14.12g, 53.9mmol) and Compound 2(a) (15.0g, 52.8mmol), followed by a degassed 2M aqueous Na_2CO_3 solution (29ml, 58.0mmol). The resultant mixture was degassed 2 times for 20-30 seconds each under a positive pressure of N_2 , followed by the addition of palladium (II) acetate (0.24g,
 20 1.1mmol) and triphenylphosphine (0.84g, 3.2mmol) and then degassed 2 more times. The reaction mixture was then refluxed under N_2 for 5h resulting in a light-yellow mixture. This mixture was cooled to 23°C resulting in the formation of a precipitate which was collected by filtration, washed successively with toluene (123ml), 2:1 MTBE/EtOAc (143ml), deionized water (123ml) and 2:1 MTBE/EtOAc (42ml) and
 25 then dried for 16h in a vacuum oven at 35°C. Yield: 18.7g (94%). R_f 0.48 (benzene). ^1H NMR: δ 0.93 (t, 3H, $J=6.80\text{Hz}$); 1.42 (m, 4H); 1.81 (m, 2H); 3.95 (s, 3H); 4.00 (t, 2H, $J=6.48\text{Hz}$); 6.97 (d, 2H, $J=8.52\text{Hz}$); 7.55 (d, 2H, $J=8.52\text{Hz}$); 7.66 (m, 6H), 8.10 (d, 2H, $J=8.20\text{Hz}$). MS(FD $^+$): m/z 374. IR(KBr): 2938, 1723 cm^{-1} . Analysis for $\text{C}_{25}\text{H}_{26}\text{O}_3$: Calcd: C, 80.18; H, 7.00; Found: C, 79.91; H, 6.94.

Preparation of Compound 4(a):**4 (a)**

- 5 A mixture of Compound 3(a) (80g, 0.21mol), 5M KOH (160ml) and cetyltrimethylammonium bromide (4.8g, 0.013mol) in xylene (800ml) was refluxed for 3h and then cooled to 10°C and filtered to provide a white solid. This solid was washed 3 times with H₂O (500ml each) to remove the catalyst and most of the base. The resultant material was treated with DME (500 ml). The pH of the solution was
- 10 adjusted to pH by the addition of 6M HCl (100ml). The resultant mixture was refluxed for 30 minutes while periodically checking the pH to assure that it remained acidic, then cooled and filtered. The resulting solid was washed successively with MTBE (400ml) and water (4x400ml) until the washings were neutral to litmus. Yield: 76 g (98% yield). ¹H NMR δ 0.89 (t, 3H, J= 6.82Hz), 1.38 (m, 4H), 1.73 (m,
- 15 2H), 3.96 (t, 2H, J= 6.3Hz), 6.95 (d, 2H, J=8.56Hz), 7.57 (d, 2H, J=8.54Hz), 7.64-7.74 (m, 6H), 8.00 (d, 2H, J=8.21Hz), 8.09 (s, 1H). MS(FD⁺) m/z 360. IR(KBr): 2958, 2937, 2872, 1688 cm⁻¹. Analysis for C₂₄H₃₄O₃: Calcd: C, 79.97; H, 6.71; Found: C, 80.50; H, 6.77.

Preparation of HOBT ester of Compound 4(a):20 A. Formation of HOBT mesylate

- To a cold (0°C) mixture of hydroxybenzotriazole hydrate (200g, 1.48mol) in anhydrous CH₂Cl₂ (1.5L), was slowly added anhydrous Et₃N (268ml, 1.92mol) while maintaining a temperature of 0-10°C, followed by the addition of methanesulfonyl chloride (126ml, 1.63mol) while maintaining a temperature of 0-5°C. The resultant
- 25 mixture was stirred for 3h at 0°C and washed successively with cold water (2 x 1.2L) and brine (1.2L). The combined organic extracts were concentrated at reduced pressure to provide a solid. This solid was recrystallized from CH₂Cl₂ (100ml) and heptane (1L). The crystals were collected by suction filtration and washed repeatedly

with a total of 1.L of heptane and then dried overnight under high vacuum (0.5 mm Hg). Yield: 245g (78%) R_f 0.55 (1:1 hexanes/ CH_2Cl_2). ^1H NMR: δ 3.58 (s, 3H), 7.46 (t, 1H, $J=7.60\text{Hz}$), 7.60 (d, 1H, $J=8.28\text{Hz}$), 7.65 (d, 1H, $J=8.56\text{Hz}$), 7.68 (d, 1H, $J=8.20\text{Hz}$), 8.05 (d, 1H, $J=8.41\text{Hz}$).

5 B. Formation of HOBT ester

A mixture of Compound 4(a) (50g, 0.14mol) and the material described above in part A (36g, 0.17mol) in DMF (650ml) was treated dropwise with Et_3N (25ml, 0.18mol), under N_2 . The resultant mixture was stirred for 4h at room temperature until all the acid was consumed, as determined by TLC (95:5 $\text{CH}_2\text{Cl}_2/\text{MeOH}$). When
10 all the acid was consumed, an aliquot of the reaction mixture (~3 pipes drops) gave a clear homogeneous solution when diluted with 3ml of 1:1 $\text{CH}_2\text{Cl}_2/\text{THF}$. The reaction mixture was then diluted with toluene (500ml), washed with water (500ml). The organic layer (containing solid product) was diluted with water (500ml) and filtered using MTBE for transferring. The solid was rinsed with MTBE (2 x 400ml) and dried
15 under vacuum to provide green-white flakes of material. NOTE: This material could be dissolved in THF and filtered to remove any remaining metal contamination. Yield: 61g (92%). R_f 0.68 (1:1 $\text{CH}_2\text{Cl}_2/\text{hexanes}$). ^1H NMR: δ 0.93 (t, 3H, $J=7.0\text{Hz}$), 1.42 (m, 4H), 1.81 (m, 2H), 4.00 (t, 2H, $J=6.53\text{Hz}$), 6.99 (d, 2H, $J=8.6\text{Hz}$), 7.42-7.59 (m, 5H), 7.71 (dd, 4H, $J=13.91\text{Hz}$, 8.40Hz), 7.86 (d, 2H, $J=8.30\text{Hz}$), 8.11 (d, 1H, $J=$
20 8.31Hz), 8.35 (d, 2H, $J=8.33\text{Hz}$). ^{13}C NMR: δ 14.03, 22.44, 28.18, 28.94, 40.10, 40.37, 68.11, 108.45, 110.11, 114.95, 118.71, 120.48, 123.04, 124.94, 124.99, 127.00, 127.23, 127.51, 127.73, 128.06, 128.82, 128.86, 131.35, 132.30, 137.15, 141.43, 143.54, 147.85, 159.15, 162.73. MS(FD $^+$): m/z 477. IR(CHCl_3): 2960, 2936, 2874, 1783, 1606 cm^{-1} . Analysis for $\text{C}_{30}\text{H}_{27}\text{N}_3\text{O}_3$: Calcd: C, 75.45; H, 5.70; N, 8.80; Found:
25 C, 75.69; H, 5.58; N, 8.92.

Preparation of Anti-fungal Compound 1(a):

Deionized water was used throughout the procedure. A mixture of Compound 5(a) (11g, 23mmol) and the nucleus of Compound 1(a) (where R is hydrogen - 92% pure by HPLC, 19.25 g, 22.2mmol) in anhydrous DMF (275ml) was stirred, under N_2
30 for 4h (until HPLC showed complete consumption of the cyclic peptide starting

material). The mixture was filtered through a bed of celite and the filtrate was concentrated under reduced pressure at 35°C to provide a paste that could be stirred. This paste was poured into MTBE (500ml) which resulted in the precipitation of a fine powder which was collected by vacuum filtration and dried to provide 27g of crude material. This material was crushed to a powder with a mortar and pestle, slurried for 5 minutes in toluene (200ml), suction filtered (slow filtered), rinsed with MTBE (100ml) and then dried *in vacuo* to provide a yellow solid. Yield: 23 g (95% pure by HPLC, retention time = 7.79 min).

Alternatively, the conversion may be carried out using an excess of the cyclic nucleus (1.1 equiv.). When the reaction is substantially complete, as indicated by HPLC, the crude material (10g of a powder) is added portion-wise to a vigorously stirred mixture of 9:1 acetone/water (60ml). Celite (2.5 g, pre-washed with a 9:1 acetone/water mixture) is added to the resultant suspension. After stirring for 2 minutes, the mixture is filtered through a bed of celite (prewashed with 9:1 acetone/water) and the cake is rinsed twice with 9:1 acetone/water (10ml). The filtrate is poured into a beaker of deionized water (200ml) while gently swirling the mixture which resulted in the formation of a precipitate. This precipitate is collected by suction filtration, rinsed with H₂O (4 x 25ml), and then dried *in vacuo* at room temperature. Yield: 6.81g (97% pure by HPLC).

The product was further purified using preparatory HPLC chromatography. R_f 0.29 (80:20 CHCl₃/MeOH). MS(FAB⁺): m/z for C₃₈H₇₄N₇O₇, Calcd: 1140.5141; Found: 1140.5103. IR(KBr): 3365, 2934, 1632, 1518 cm⁻¹.

Preparation of a fructose complex with Compound 1(a):

A jacketed reactor was charged with 1 equivalent of Compound 1(a), 8 equivalents of fructose and a sufficient quantity of methanol to make 58 mg/ml of Compound 1(a). The mixture was heated to 50-55°C until the dissolution was complete. The solution was cooled to 45°C. After seeding at 45°C, the seeded solution was cooled to 25°C at a cooling rate of -2 degrees/hour. The mixture was further cooled to 0°C over 2 hours (cooling rate = -12.5 degrees/hour) and then stirred at 0°C for 12 hours. The product was isolated by vacuum filtration, washed with cold

methanol containing 1% fructose on a weight/weight basis, and then dried 24 hours in a 30°C vacuum oven. Assays were performed on a gradient HPLC system equipped with a 15 cm x 4.6 mm, 3.5 micron particle size Zorbax™ SB-C18 or XDB-C18 analytical column.

Pharmaceutical Formulations

The following Examples illustrate formulations prepared by the processes of the invention. The Examples are not intended to limit the scope of the invention in any respect and should not be so construed.

The quantity of Compound 1(a) was determined by calculating the theoretical potency needed for the experiments and dividing that value by the "as-is" HPLC potency of the compound.

Example 1

Example 1 illustrates *in situ* formation of an echinocandin/carbohydrate complex using a fluid bed spray process.

Solution A was prepared by adding 523.70 g of fructose to 760.0 ml of warm purified water while stirring. The solution was then cooled to room temperature.

Solution B was prepared by adding 2,850.0 ml of acetone with stirring to 950.0 ml of purified water cooled to 20-25°C. With stirring, 523.70 g of Compound 1(a) was slowly added to the acetone/water solution.

Solution A was added slowly to solution B with stirring. The clear solution was then sprayed onto a fluidized bed of granular mannitol 2080 (available from SPI Polyols, New Castle, DE) that had been added to the dryer bowl of a Glatt GPCG-3 fluid bed granulation unit (available from Glatt Air Techniques, Inc., Ramsey, NJ). The initial inlet temperature of the unit was set at 30°C and then increased to 45°C after the spraying was complete to remove the excess solvent. After drying, the bags were shaken and the contents transferred to a plastic lined container.

The process provided 3,332.7 g (90.9% yield) of a free flowing granular material having a fine texture. Karl Fischer analysis = <1% moisture. Bulk density:

- 0.67 g/ml Tap density after 300 taps of the material = 0.79 g/ml. Relative stability of the sprayed granules was determined as the relative ratio of the main degradation products of the control (i.e., bulk Compound 1(a)) versus the test material (i.e., sprayed granules). The degradation products are determined by high performance liquid chromatography (HPLC) and recorded as percent total related substances (%TRS). After storing the sprayed granules in a closed high density polyethylene brown opaque bottle for 1 week at 40°C, the sprayed granules exhibited a lower %TRS (6.50% TRS) than the bulk Compound 1(a) (7.80% TRS) thus indicating an enhancement in thermal stability for the sprayed granules.
- 10 After two weeks of storage at 40°C, the sprayed granules continued to exhibit a lower %TRS than the bulk Compound 1(a) (7.42% TRS for the sprayed granules versus 8.79% TRS for Compound 1(a)).